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Key indicators

Single-crystal X-ray study

T = 180 K

Mean $\sigma(\text{C}-\text{C}) = 0.002 \text{ \AA}$

R factor = 0.045

wR factor = 0.134

Data-to-parameter ratio = 27.3

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

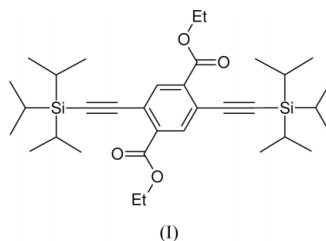
Diethyl 2,5-bis(triisopropylsilylethynyl)-terephthalate

In the crystal structure of the title compound, $\text{C}_{34}\text{H}_{54}\text{O}_4\text{Si}_2$, at 180 K, the molecules lie on inversion centres.

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Experimental

The title compound was prepared according to a procedure similar to that for the related compound diethyl 2,5-bis(1-decynyl)-terephthalate (Zhang *et al.*, 2000). Crystals suitable for X-ray analysis were prepared by slow evaporation of a dichloromethane solution.

Crystal data

$\text{C}_{34}\text{H}_{54}\text{O}_4\text{Si}_2$
 $M_r = 582.95$
 Triclinic, $P\bar{1}$
 $a = 7.6344 (4) \text{ \AA}$
 $b = 7.7017 (4) \text{ \AA}$
 $c = 16.2680 (9) \text{ \AA}$
 $\alpha = 89.224 (2)^\circ$
 $\beta = 77.873 (2)^\circ$
 $\gamma = 69.352 (1)^\circ$
 $V = 873.19 (8) \text{ \AA}^3$

$Z = 1$
 $D_x = 1.109 \text{ Mg m}^{-3}$
 Mo $K\alpha$ radiation
 Cell parameters from 5107 reflections
 $\theta = 2.6\text{--}30.5^\circ$
 $\mu = 0.14 \text{ mm}^{-1}$
 $T = 180 (2) \text{ K}$
 Block, colourless
 $0.35 \times 0.20 \times 0.20 \text{ mm}$

Data collection

Bruker–Nonius X8APEX-II CCD diffractometer
 Thin-slice ω and φ scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 2003)
 $T_{\min} = 0.800$, $T_{\max} = 0.974$
 11772 measured reflections

5139 independent reflections
 4112 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.025$
 $\theta_{\text{max}} = 30.5^\circ$
 $h = -10 \rightarrow 10$
 $k = -10 \rightarrow 10$
 $l = -20 \rightarrow 23$

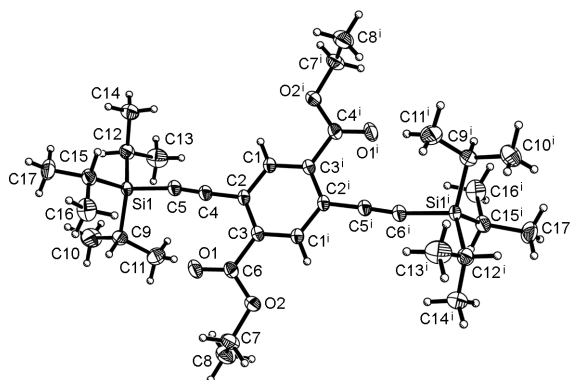


Figure 1

The molecular unit of (I), showing displacement ellipsoids at the 50% probability level for non-H atoms. H atoms are shown as spheres of arbitrary radius. [Symmetry code (i): $1 - x, 1 - y, 1 - z$.]

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.045$
 $wR(F^2) = 0.134$
 $S = 1.07$
 5139 reflections
 188 parameters
 H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0768P)^2 + 0.1418P]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.47 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.23 \text{ e } \text{\AA}^{-3}$

H atoms bound to C atoms were positioned geometrically and allowed to ride during subsequent refinement with C–H = 0.95 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for the H atom bound to C1; C–H = 0.99 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for H atoms bound to C7; C–H = 1.00 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for H atoms bound to C9, C12 and C15; C–H = 0.98 Å and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for H atoms of the methyl groups. The methyl groups were allowed to rotate about their local threefold axes.

Data collection: *APEX2* (Bruker–Nonius, 2003); cell refinement: *SAINT* (Bruker, 2003); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2000); program(s) used to

refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

The structure is the first to be produced from a new Bruker–Nonius X8APEX-II system. We are grateful to the Danish Natural Science Research Council (SNF) and Calsbergfondet for providing the funds to purchase this equipment.

References

- Bruker (2003). *SAINT*. Version 7.06a. Bruker AXS Inc., Madison, Wisconsin, USA.
 Bruker–Nonius (2003). *APEX2*. Version 1.0–8. Bruker–Nonius BV, Delft, The Netherlands.
 Sheldrick, G. M. (2003). *SADABS*. Version 2.10. Bruker AXS Inc., Madison, Wisconsin, USA.
 Sheldrick, G. M. (2000). *SHELXTL*. Version 6.10. Bruker AXS Inc., Madison, Wisconsin, USA.
 Zhang, Q., Shi, C., Zhang, H.-R. & Wang, K. K. (2000). *J. Org. Chem.* **65**, 7977–7983.